

# REPORT DOCUMENTATION PAGE

Form Approved  
OMB No. 0704-0188

The public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing the burden, to the Department of Defense, Executive Service and Communications Directorate (0704-0188). Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number.

PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ORGANIZATION.

1. REPORT DATE (DD-MM-YYYY) 08-01-2012			2. REPORT TYPE Abstracts and Oral Presentations		3. DATES COVERED (From - To)	
4. TITLE AND SUBTITLE Abstracts and Oral Presentations of: 16th Int. Congress of Marine Corrosion & Fouling (ICMCF) June 24-28, 2012 in Seattle, WA					5a. CONTRACT NUMBER	
					5b. GRANT NUMBER	
					5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S)					5d. PROJECT NUMBER	
					5e. TASK NUMBER	
					5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Naval Research Laboratory Oceanography Division Stennis Space Center, MS 39529-5004					8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES) Office of Naval Research One Liberty Center 875 North Randolph Street, Suite 1425 Arlington, VA 22203-1995					10. SPONSOR/MONITOR'S ACRONYM(S) ONR	
					11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release, distribution is unlimited.						
13. SUPPLEMENTARY NOTES						
14. ABSTRACT						
15. SUBJECT TERMS						
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT UU	18. NUMBER OF PAGES	19a. NAME OF RESPONSIBLE PERSON	
a. REPORT Unclassified	b. ABSTRACT Unclassified	c. THIS PAGE Unclassified			19b. TELEPHONE NUMBER (Include area code)	

## Sulfide Production and Corrosion in Seawater During Exposure to FAME Alternative Fuel



**Jason S. Lee**

Richard I. Ray

Brenda J. Little

Naval Research Laboratory  
Stennis Space Center, MS



Kathleen E. Duncan

Athenia L. Oldham

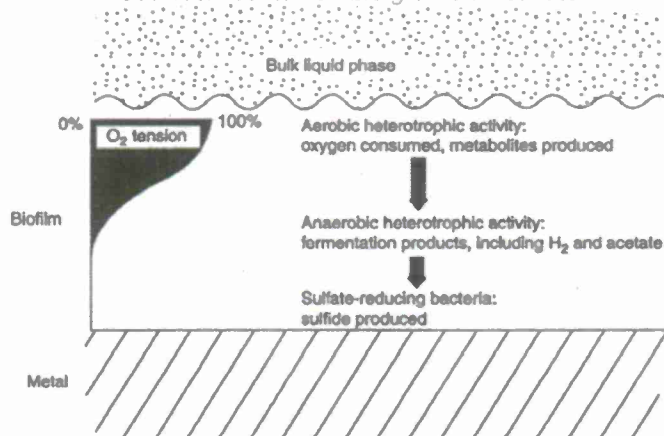
Irene A. Davidova

Joseph M. Suflita

University of Oklahoma  
Norman, OK

## Sulfide Derivitization

Seawater contains  $\sim 2.0$  grams  $L^{-1}$  sulfate



20121203012

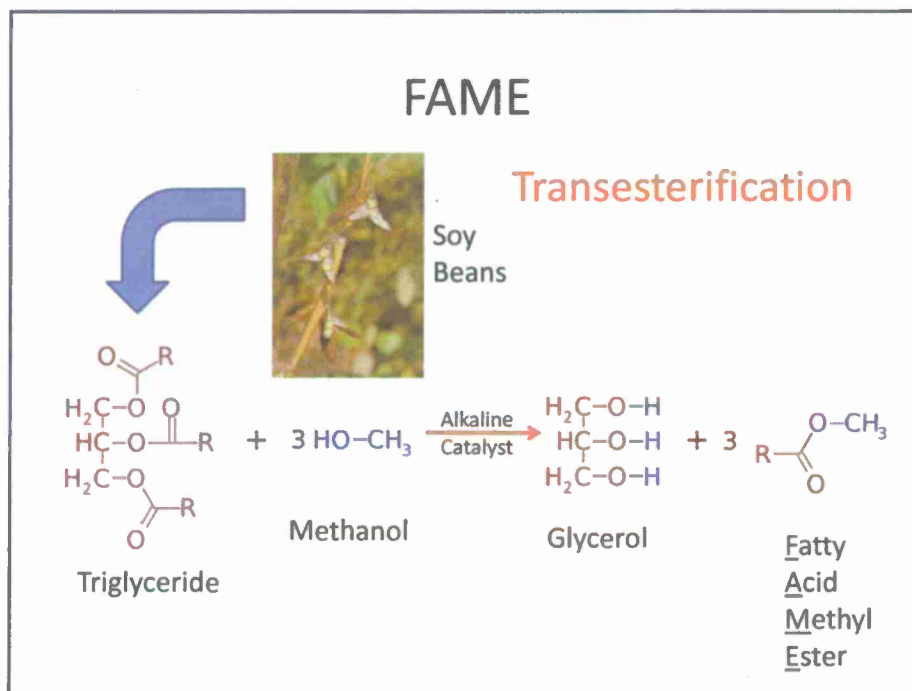
## Sulfate-Reducing Bacteria



## Initial Seawater Chemistries

Seawaters	pH	Salinity (g/L)	Total Organic Carbon (mg/L)	Sulfate (mg/L)
Key West	7.82	38	1.79	3864
Persian Gulf	7.98	44	1.94	4696

## FAME



## Exposure Chamber

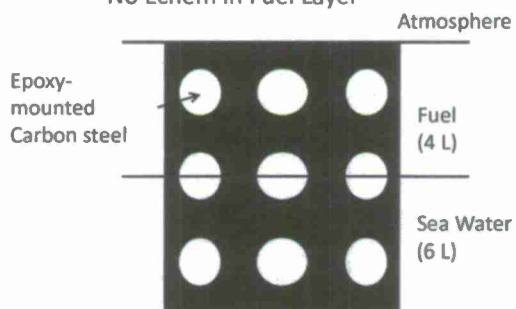
### Anaerobic Chamber

- bal. N<sub>2</sub>, 10% H<sub>2</sub>, 0.1% CO<sub>2</sub> - maintain pH ~8

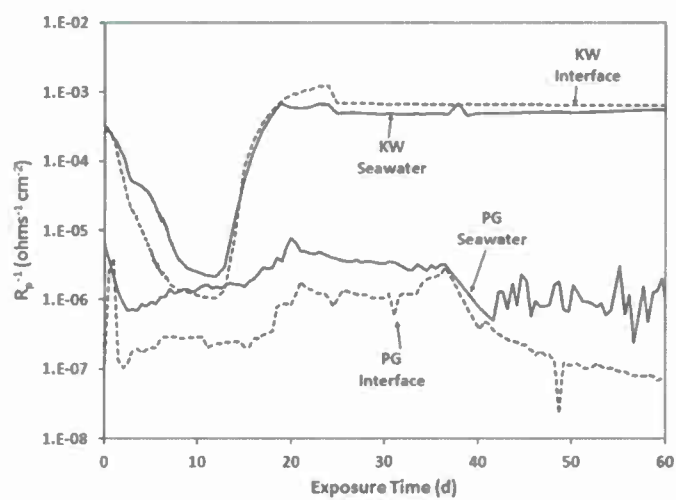
Polarization Resistance (R<sub>p</sub>)

Corrosion Potential (E<sub>corr</sub>)

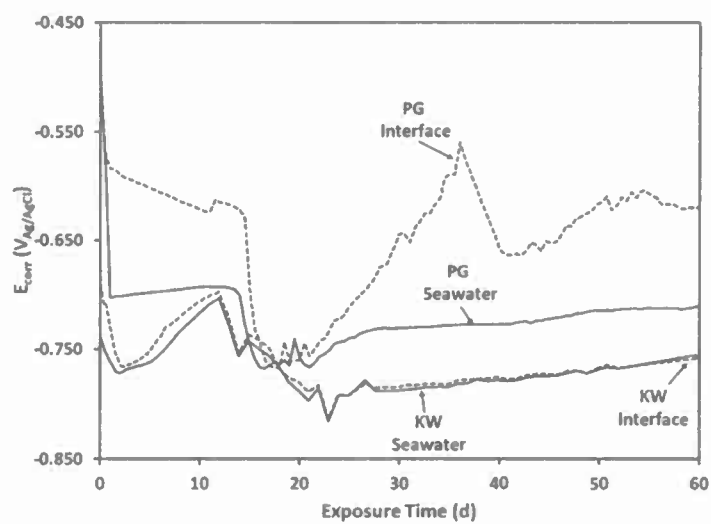
- No Echem in Fuel Layer



### Corrosion Rate ( $R_p^{-1}$ )

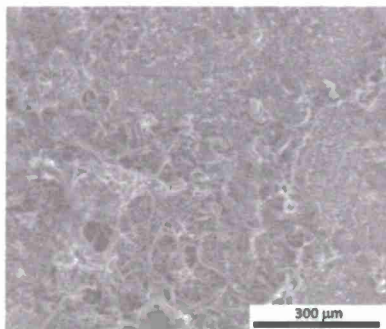


### Corrosion Potential ( $E_{\text{corr}}$ )

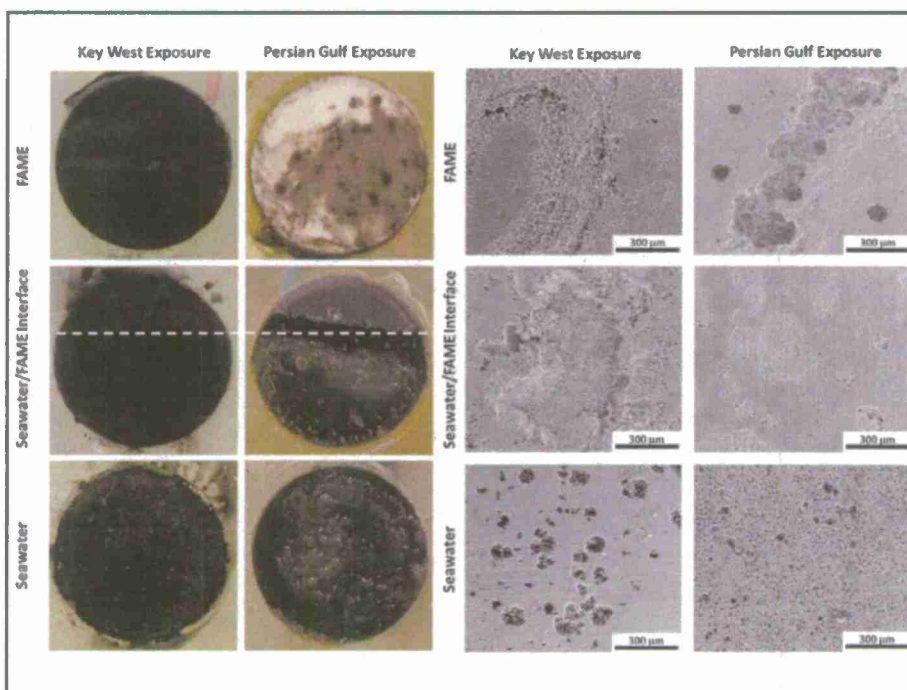
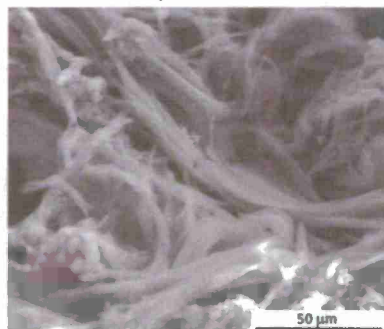


## Fouling at the Interface

Persian Gulf



Key West



## Corrosion Product Analysis

Energy Dispersive Spectroscopy

Electrode Position	Sulfur (wt%)		Chlorine (wt%)	
	KW	PG	KW	PG
FAME	19.7	1.7	4.5	0.49
FAME/SW Interface	23.6	0.5	4.7	0
Seawater	30.2	2.4	6.6	0

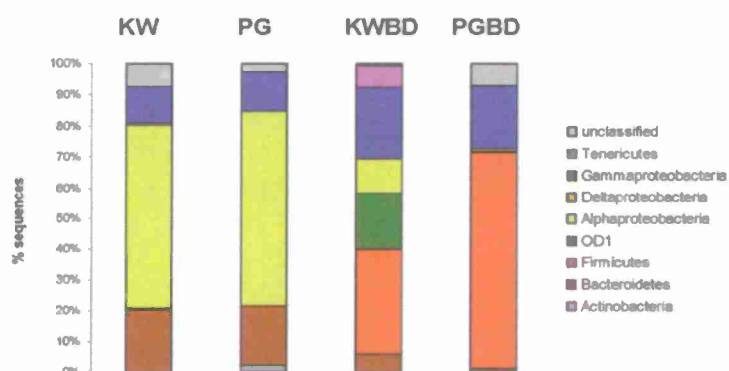
## Sulfate reduction activity (SRA)

Sample	SRA $\mu\text{mol S / L/day}$	
	Persian Gulf Seawater PG	Key West Seawater KW
in situ (no additions)	$11.96 \pm 1.33$	$17.7 \pm 3.3$
Amended with lactate	$23.5 \pm 1.7$	115
Amended with crude oil*	$10.3 \pm 2.3$	$13.95 \pm 0.75$
Amended with crude oil and inoculated with strain Lake**	$155 \pm 6.7$	$264 \pm 40$
Sterile Control	$7.95 \pm 1.7$	$7.5 \pm 3.5$

\* sterile crude oil

\*\* *Desulfogloebe* strain Lake, an alkane-degrading sulphate-reducing bacterium

## Bacterial 16rRNA Pyrosequencing



## Quantitative PCR

Estimates from qPCR	KW*	PG	KWBD	PGBD
Bacterial cells/ml	$2.75 \times 10^7$	$2.66 \times 10^7$	$4.97 \times 10^5$	$1.72 \times 10^5$
Dsr-bearing cells/ml**	3.17	BDL	BDL	BDL
Aps-bearing cells/ml***	BDL	BDL	BDL	BDL
Archaeal cells/ml	$3.05 \times 10^3$	$2.19 \times 10^3$	BDL	BDL
Mcr-bearing cells/ml****	$2.48 \times 10^3$	$2.48 \times 10^0$	$1.21 \times 10^2$	$4.74 \times 10^1$

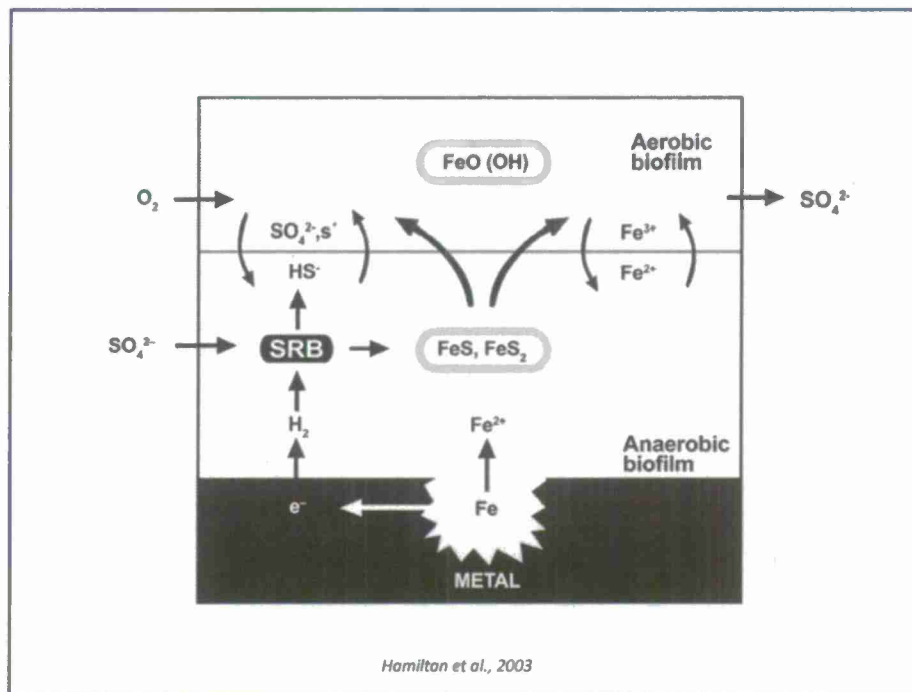
\*KW: Key West seawater; PG: Persian Gulf seawater; KWBD: FAME diesel incubated with KW seawater; PGBD: FAME diesel incubated with PG seawater.

\*\* Dsr-bearing cells: cells that contain a copy of the gene coding for dissimilatory (bi)sulphite reductase, e.g. SRB.

\*\*\* Aps-bearing cells: cells that contain a copy of the gene coding for adenosine-5'-phosphosulphate reductase, e.g. SRB.

\*\*\*\* Mcr-bearing cells: cells that contain a copy of the gene coding for subunit *a* of methyl-S-CoM methylreductase, e.g. methanogens.





## Conclusions

- Sulfide influenced corrosion rates of carbon steel exposed to seawaters and FAME diesel did not correlate with initial concentrations of sulfate, chloride or organic carbon in the seawater. KW >> PG
- A microbial community developed with low numbers of SRB after seawater was incubated with the alternative fuel
- Significantly higher elemental concentrations of sulfur and chlorine were detected in corrosion products in the KW exposure compared to PG
- Initially higher estimates of Dsr- and Mcr-bearing cells (i.e., SRB and methanogens) in KW compared with PG provide the only indication that KW seawater will support more sulfate reduction.
- The inability to predict corrosivity of particular seawaters from a limited set of chemical and microbial parameters demonstrates that simple models in which SRB abundance is directly associated with rate or extent of corrosion are inadequate.

## MIC-5

## MARINE CORROSION IN FUEL SYSTEMS

Brenda J. Little<sup>1</sup>, Jason S. Lee<sup>1</sup>, Richard I. Ray<sup>1</sup>, Deniz F. Aktos<sup>2</sup>, Kathleen E. Duncon<sup>2</sup>, and Joseph M. Suflita<sup>2</sup>

The relationship between corrosion and biodegradation of bio- and petroleum-based fuels exposed to seawater is being evaluated. To date the fuels have included petroleum diesel (F76) and jet propellant (JP) 5, hydroprocessed (HP) bio-based lipids from renewable stocks (e.g. camelina and algae) and blends. Experiments have been conducted with aerobic seawater and unprotected carbon steel coupons under stagnant conditions. i.e., there were no attempts to influence the distribution or concentration of oxygen in the sealed vessels. In all cases the dissolved oxygen (DO) in the seawater was below the detection limits of the DO probe (100 ppb) within a few days of incubation, independent of fuel composition. Corrosion was due to microbiologically produced sulfides reacting with the carbon steel. There were few differences in electrochemically measured corrosion rates in incubations amended with any of the fuels or their blends. In the experiments that have been examined in detail, transient oxygen influenced the microbial biodegradation of fuels and resulted in a suite of characteristic metabolites. Detection of catechols confirmed the exposure of incubations to oxygen. Clone library analysis indicated higher proportions of Firmicutes, Deltaproteobacteria (primarily sulfate-reducing bacteria), Chloroflexi, and Lentisphaerae in incubations exposed to fuels than the original seawater. Relative proportions of sequences affiliated with these bacterial groups varied with fuel. Methanogen sequences similar to those of *Methanobrevibacter* were also found in multiple incubations. Despite the dominance of characteristically anaerobic taxa, sequences coding for an alkane monooxygenase from marine hydrocarbon-degrading genera was observed, suggesting that organisms with this metabolic potential survived the incubation. The current hypothesis is that initial aerobic oxidation of fuel components resulted in the formation of a series of intermediates that were used by anaerobic seawater microbial communities to support their metabolism, sulfide production, and carbon steel microbiologically influenced corrosion. The more precise relationship between oxygen, microbial activity and corrosion is underway with more precise DO probes (4 ppb resolution).

## MIC-6

## SULFIDE PRODUCTION AND CORROSION IN SEAWATER DURING EXPOSURE TO FAME ALTERNATIVE FUEL

Jason S. Lee, Richard I. Ray, Brenda J. Little

Naval Research Laboratory, Stennis Space Center, MS 39529

Kathleen E. Duncan, Athenia L. Oldham, Irene A. Davidova and Joseph M. Suflita

Department of Botany and Microbiology and Institute for Energy and the Environment, University of Oklahoma, Norman, OK, 73019

Experiments were designed to evaluate corrosion-related consequences of storing/transporting fatty acid methyl ester (FAME) alternative diesel fuel in contact with natural seawater under anaerobic conditions. Coastal Key West, FL, and Persian Gulf seawaters, representing an oligotrophic and a more organic- and inorganic mineral-rich microbial coastal seawater environment, respectively, were used in 60-day studies with unprotected carbon steel. Despite low numbers of sulfate reducing bacteria in the original waters and after FAME diesel exposure, sulfide levels and corrosion increased markedly due to microbial sulfide production. The original microflora of the two seawaters was similar with respect to major taxonomic groups but with markedly different species. After exposure to FAME diesel the microflora of both waters changed dramatically, with Clostridiales (Firmicutes) becoming dominant. Microbial sulfide production was stimulated in both seawaters by the presence of FAME.

